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February 9, 2016 Igcse  
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Notes 1 Sulphuric Acid:  
It is a colourless oily yellow liquid, with density slightly higher than water. Having a boiling point of  $338^{\circ}\text{C}$ . It is soluble in water and emits heat when dissolved in water - exothermic in nature.

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$2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O} + 4\text{e}^-$  -->  
4oh- If we combine the  
ionic equations, we still  
get water as product of  
hydrogen and oxygen,  
but the energy  
produced is now  
electrical energy:  $2\text{H}_2$   
(g) +  $\text{O}_2$  (g) -->  $\text{H}_2\text{O}$   
(g) + electrical energy

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The shells fill in order,  
from lowest energy  
level to highest energy  
level • All the elements  
in a group have the



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same number of  
electrons in their outer  
... O . O . H H + . • • •  
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$C_6H_{12}O_6(aq) \rightarrow 2C_2H_5OH(aq) + 2CO_2(g)$  Temperature is kept constant at 37 °C to prevent destruction of yeast at higher temperatures. Oxygen is removed by

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limewater and carbon dioxide is produced during fermentation. Alcohol is separated from solution by fractional distillation. 2. Reacting Ethene with Steam

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