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Using the Limiting Reactants Gizmo™, you can determine which reactant is limiting in various scenarios. To begin, make sure $\text{H}_2 + \text{O}_2$ becomes H_2O is selected. The small “2” in H_2 , O_2 , and H_2O is a subscript. Subscripts represent the number of atoms in a molecule.

Student Exploration: Limiting Reactants (ANSWER KEY)

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Explore the concepts of limiting reactants, excess reactants, and theoretical yield in a chemical reaction. Select one of two different reactions, choose the number of molecules of each reactant, and then

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observe the products created and the reactants left over. Launch Gizmo. Limiting Reactants.

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Q1 Q2 Q3 Q4 Q5

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Chemistry 2B and 4B

5/5 Questions &

Answers 1. The limiting

reactant in a chemical

reaction is the reactant

that A. is used up at

the end of the reaction.

B. is left over after the

reaction. C. has the

largest number of

atoms. D. has the

fewest number of

atoms. Explanation:

The limiting reactant

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will be the reactant that is "fully consumed" upon completion of the ...

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Explore the concepts of limiting reactants, excess reactants, and theoretical yield in a chemical reaction.

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Select one of two different reactions, choose the number of molecules of each reactant, and then observe the products created and the reactants left over.

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Sometimes during a chemical reaction, one type of reactant will be used up before the other reactants. This

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reactant is the limiting reactant. Using the Limiting Reactants Gizmo™, you can determine which reactant is limiting in various scenarios. To begin, make sure $H_2 + O_2 \rightarrow H_2O$ is selected.

Student Exploration: Limiting Reactants

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Answer Key Using the Limiting Reactants Gizmo™, you can determine which reactant is limiting in various scenarios. To begin, make sure $H_2 + O_2$ becomes H_2O is selected. The small “2” in H_2 , O_2 , and H_2O is a subscript.

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Answer Key

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Name: ___Samuel

Chen ___ Date: _____

Student Exploration:

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Vocabulary: chemical
equation, chemical

formula, chemical

reaction, coefficient,

limiting reactant,

molecule, product,

reactant, subscript

Prior Knowledge

Questions (Do these

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BEFORE using the Gizmo.) Imagine you and your friends are making hot dogs.

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Name_Samuel Chen

Date Student ...

in a molecule page 5
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determine which

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reactant is limiting in various scenarios to begin make sure h₂o₂ becomes h₂o is selected the small 2 in h₂o₂ and h₂o is a subscript

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Limits the extent of the reaction and, thereby, determines the amount of product formed [Is completely consumed in a chemical reaction]

Theoretical Yield. The maximum amount of product that can be produced from a given amount of reactant.
Reassemble.

**Chem. - Limiting
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Flashcards | Quizlet

$\Rightarrow 0.944 \text{ g CaCO}_3 \rightarrow$
Limiting Reagent . 2.00
 $\text{g CaCl}_2 \times (1 \text{ mol} /$
 $108.98 \text{ g}) \times (1 \text{ mol}$
 $\text{CaCO}_3 / 1 \text{ mol CaCl}_2)$
 $\times (100.08 \text{ g CaCO}_3 / 1$
 $\text{mol}) \Rightarrow 1.834 \text{ g CaCO}_3$
 \rightarrow Excess Reagent.

Which reactant is there
an excess of? There is
a excess of CaCl_2 .

Where can it be found?
It can be found in the
liquid in the flask.

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