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A covalent compound made of carbon and hydrogen atoms.

hydrocarbons chapter 22 Flashcards and Study Sets | Quizlet

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Two polar compounds will form a solution and two non-polar compounds will be able to dissolve together. However, polar will not mix with non-polar, e.g.: oil (non-polar) and water (polar).
Work Step by Step The rule is that "like dissolves like."

Chapter 22 - Hydrocarbon Compounds - 22.1 Hydrocarbons ...

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Hydrocarbon.

Hydrocarbons are

compounds comprised exclusively of carbon and hydrogen and they are by far the dominant components of crude oil, processed petroleum

hydrocarbons

(gasoline, diesel, kerosene, fuel oil, and lubricating oil), coal tar, creosote, dyestuff, and pyrolysis waste products. From:

Treatise on

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Geochemistry, 2007.
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Organic compounds that contain the maximum number of hydrogen atoms per carbon atoms are called _____ compounds. 3. Which family of hydrocarbons

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are always saturated
compounds? 4.

Circle the letter of the
correct name for the
alkene shown below. a.
2,3-dimethyl-3-pentene
c.
2,3-dimethyl-2-pentene

SECTION 22.1 HYDR OCARBONS(pages 693-701)

unsaturated
hydrocarbonsA
molecule where some
of the carbons are
connected to fewer

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than four neighboring atoms. Usually the molecule contains double or triple bonds. Hydrocarbons are the simplest class of organic compounds and are composed solely of hydrogen and carbon. This class can be further divided into two groups: aliphatic hydrocarbons and aromatic hydrocarbons.

Introduction to
Page 10/27

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Worksheet 7 Naming
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Naming Ionic
Compounds Worksheet
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Worksheet 7 Naming
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Hydrocarbon
Compounds Guide
Answers A controlled
process by which
hydrocarbons are
broken down or
rearranged into
smaller, more useful
molecules natural gas
This is typically made
up of 80% methane,
10% ethane, 4%
propane, and 2%
butane Study Guide:

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Chapter 22

Hydrocarbon

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Hydrocarbons are the simplest organic compounds, but they have interesting physiological effects. These effects depend on the size of the hydrocarbon molecules and where on or in the body they are applied.

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Alkanes of low molar mass—those with from 1 to approximately 10 or so carbon atoms—are gases or light liquids that act as anesthetics.

Chapter 1 - Organic Chemistry Review / Hydrocarbons - CHE

...

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Hydrocarbons are
organic molecules that
only contain two

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elements: carbon and hydrogen. Common hydrocarbons include methane, propane, and butane.

Section 22hydrocarbon Compound Answer

Hydrocarbon
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Answers A controlled
process by which
hydrocarbons are
broken down or
rearranged into
smaller, more useful

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molecules natural gas
This is typically made
up of 80% methane,
10% ethane, 4%
propane,

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compounds that
provide a source of
energy and raw
materials. Real-World

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Reading Link If you have ridden in a car or a bus, you have used hydrocarbons. The gasoline and diesel fuel that are used in cars, trucks, and buses are hydrocarbons. ...

Study Guide For Hydrocarbons Answers

Hydrocarbons Study Guide Answers A more appropriate definition of organic chemistry is: The chemistry of

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hydrocarbons
(compounds of carbon and hydrogen only) and the derivatives of hydrocarbons, (compounds containing C, H, and O, N, S, P, halogens etc). In some organic compounds, the metals such as Na, K, Ca and many others can be found.

Organic Chemistry Hydrocarbons Study Guide Answers

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hydrogenation reaction
is possible only for
compounds with
double or triple bonds,
and such compounds
are said to be
unsaturated
hydrocarbons.

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of the hydrocarbon
molecules shown
below: 1) 22 2)

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